

1,2-Bis(diphenylphosphino)-1,2-dicarba-*closo*-dodecaboraneDao-Peng Zhang, Jian-Min Dou,*
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The title compound, $C_{26}H_{30}B_{10}P_2$, was synthesized by the reaction of 1,2-dilithiocarborane with diphenylchlorophosphine in diethyl ether. The molecular symmetry of the molecule deviates from C_{2v} because of the different mutual orientations of the phenyl groups. The two P atoms of the $-PPh_2$ groups and two cage C atoms are almost coplanar, with a $P-C-C$ torsion angle of $10.6(3)^\circ$.

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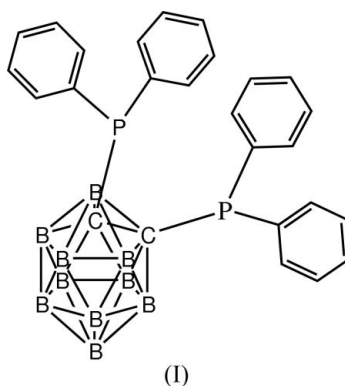
Key indicators

Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(C-C) = 0.006$ Å
 R factor = 0.049
 wR factor = 0.145
Data-to-parameter ratio = 19.6

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Comment

Carboranes, despite their high cost, are uniquely suitable for several applications (Plešek, 1992). Owing to their interesting chemical and physical properties, organic or organo-element derivatives of dicarba-*closo*-dodecaboranes have received much attention during the past few decades. These types of compounds can be used as catalysts (Hart & Owen, 1985; Longato & Bresadola, 1982), as precursors for ceramic materials (Hsu *et al.*, 1991) and in medical areas (Hawthorne, 1993; Gielen *et al.*, 1995). 1,2-Dicarba-*closo*-dodecaborane is an icosahedral cluster with two C atoms in adjacent positions. The first diphosphine derivative was reported by Alexander & Schroeder (1963). Since then, these types of compounds have been widely used as ligands in the area of transition metal chemistry (Al-Baker *et al.*, 1987; Crespo *et al.*, 1992; Bembenek *et al.*, 1994).



The molecular structure of the title compound is shown in Fig. 1. The molecule is composed of an icosahedral carborane skeleton with $-PPh_2$ groups bonded to the two adjacent cage C (Cc) atoms. The relative orientations of the phenyl groups are quite different, and, as a result, the molecular symmetry of the molecule deviates from C_{2v} . The two phosphorus and two cage C atoms are almost coplanar, the $P1-C1-C2-P2$ torsion angle being $10.6(3)^\circ$. The two $P-Cc-Cc$ angles in the title compound are $116.6(2)$ and $111.07(19)^\circ$ for $C2-C1-P1$ and $C1-C2-P2$, respectively. The corresponding angles in 1,2-

(PClPh)₂-1,2-C₂B₁₀H₁₀ (Balema *et al.*, 1998) are 112.89 (13) and 112.19 (13)°, and in 1,2-(PⁱPr₂)₂-1,2-C₂B₁₀H₁₀ (Kivekäs *et al.*, 1995) they are 112.9 (2) and 112.3 (1)°. The Cc–P, Cc–B, B–B and Cc–Cc distances in the title compound (Table 1) are in agreement with these two literature structures.

Experimental

The title compound was prepared according to the literature method (Alexander & Schroeder, 1963). The product was recrystallized from petroleum ether (b.p. 333–383 K) and dried in a vacuum (65%) (m.p. 490–493 K). The white solid was dissolved in dichloromethane, and crystals suitable for X-ray diffraction were obtained after partial evaporation of the solvent.

Crystal data

C ₂₆ H ₃₀ B ₁₀ P ₂	$D_x = 1.218 \text{ Mg m}^{-3}$
$M_r = 512.54$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 4735 reflections
$a = 16.315 (6) \text{ \AA}$	$\theta = 2.3\text{--}26.3^\circ$
$b = 10.387 (4) \text{ \AA}$	$\mu = 0.17 \text{ mm}^{-1}$
$c = 17.768 (6) \text{ \AA}$	$T = 298 (2) \text{ K}$
$\beta = 111.845 (5)^\circ$	Block, white
$V = 2795.0 (17) \text{ \AA}^3$	$0.43 \times 0.41 \times 0.40 \text{ mm}$
$Z = 4$	

Data collection

Bruker SMART CCD area-detector diffractometer	6721 independent reflections
φ and ω scans	4367 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.046$
$T_{\text{min}} = 0.930$, $T_{\text{max}} = 0.934$	$\theta_{\text{max}} = 28.4^\circ$
17791 measured reflections	$h = -21 \rightarrow 21$
	$k = -12 \rightarrow 13$
	$l = -11 \rightarrow 23$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.065P)^2 + 1.0346P]$
$R[F^2 > 2\sigma(F^2)] = 0.049$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.145$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.58 \text{ e \AA}^{-3}$
6721 reflections	$\Delta\rho_{\text{min}} = -0.33 \text{ e \AA}^{-3}$
343 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

P1–C1	1.889 (3)	B3–C1	1.717 (5)
P2–C2	1.880 (3)	B3–B9	1.778 (5)
B6–C2	1.708 (5)	B3–B4	1.792 (6)
B6–C1	1.732 (5)	B4–C1	1.711 (5)
B6–B11	1.763 (5)	B4–B9	1.774 (6)
B6–B7	1.779 (5)	B4–B10	1.775 (6)
B6–B5	1.783 (5)	B4–B5	1.778 (6)
B7–C2	1.714 (5)	B5–C1	1.714 (5)
B7–B11	1.764 (6)	B5–B10	1.778 (6)
B7–B12	1.767 (6)	B5–B11	1.782 (6)
B7–B8	1.772 (6)	B11–B12	1.789 (6)
B8–C2	1.716 (5)	B11–B10	1.799 (6)
B8–B12	1.768 (6)	B12–B10	1.767 (6)
B8–B9	1.775 (6)	B12–B9	1.776 (6)
B8–B3	1.779 (5)	B9–B10	1.779 (6)
B3–C2	1.714 (5)	C2–C1	1.697 (4)
C1–C2–P2	111.07 (19)	C2–C1–P1	116.6 (2)

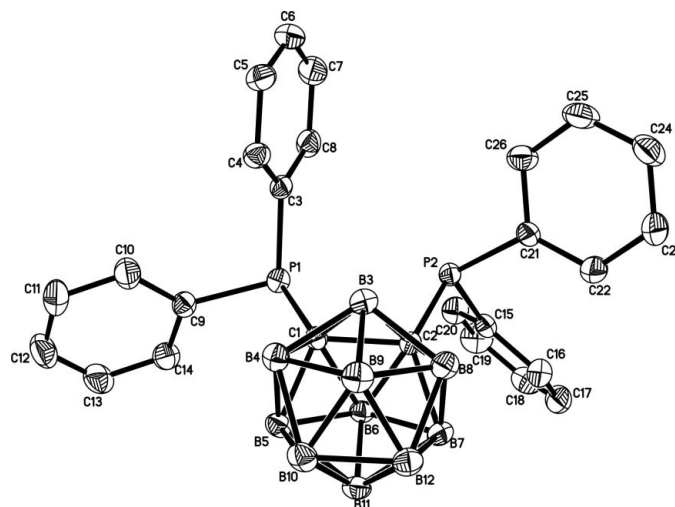


Figure 1

The molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level. H atoms have been omitted for clarity.

All H were placed in calculated positions with C–H = 0.93 Å and B–H = 1.10 Å. They were included in the refinement with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{B})$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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